Hall Ticket Number :						R-19
Code: 19A511T						K-17

I B.Tech. I Semester Supplementary Examinations July 2022

Problem Solving and C Programming

(Common to All Branches)

Max. Marks: 70 Time: 3 Hours Answer any five full questions by choosing one question from each unit (5x14 = 70 Marks)Marks **UNIT-I** 1. a) What is Programming Language? What is the generation of programming Language? 10M Describe it briefly. b) Describe Structure of C Language 4M OR 2. a) Distinguish between printf() and scanf() in the C language 7M b) What is a flow chart? How it is different from an Algorithm 7M UNIT-II 3. a) Explain with examples, any Four types of operators. 7M Explain for loop and nested for loop with suitable example. 7M OR Describe Conditional Statements Used in C Language 7M b) Write a program on calculating area and perimeter of square 7M **UNIT-III** 5. a) What is a function in c? How function is declared. Explain with an example. 7M Illustrate the storage classes extern, static and auto with an example. 7M OR 6. a) Explain the following string handling functions with examples: (i) strcpy() (ii) strcat() (iii) strrev() (iv) strlen 8M b) Explain Preprocessor commands with examples. 6M **UNIT-IV** 7. a) What is a pointer? What are the advantages of pointers? 7M Explain dynamic memory allocation with examples 7M

UNIT-V

Write a c program to swap two numbers using call by value and call by reference.

9. a) Define Structure? How structures are initialized? Explain with example. 7M

OR

b) Write a C program read and write the content of the file using fprintf() and fscanf() functions.

OR

10. a) Explain the following functions in files:

(i) fseek() (ii) ftell() (iii) foef() (iv) fopen()

8. a) Explain about pointer arithmetic and arrays with example.

b) Define and write the syntax of the structure and union and give example for each one

7M

7M

7M

8M

7M

	Hall Ticket Number :			
		R-1	9	
	Code: 19AC11T I B.Tech. I Semester Supplementary Examinations Jul Algebra and Calculus (Common to All Branches) Max. Marks: 70 Answer any five full questions by choosing one question from each unit	Time: 3		_
	******	Marks	СО	Blooms Level
1.	Solve the system of equations by matrix method $x + y + z = 6, 2x + 3y - 2z = 2, 5x + y + 2z = 13$ OR	14M	CO1	L3
2.				
0	UNIT-II	14M	CO1	L3
3.	Verify Cayley-Hamilton theorem for the matrix $A = \begin{bmatrix} 1 & 2 & 3 \\ 2 & 4 & 5 \\ 3 & 5 & 6 \end{bmatrix}$ and hence A^{-1} using Cayley-Hamilton theorem.	find 14M	CO2	L2
 4. 5. 	using orthogonal transformation. UNIT-III	by 14M	CO2	L3
J.	If $u = x^2 - y^2$, $v = 2xy$ where $x = r\cos_{\pi}$, $y = r\sin_{\pi}$, then show that $\frac{\partial(u,v)}{\partial(r,_{\pi})} = 4r^3$ OR	14M	CO3	L2
6.		t is 14M	CO3	L3
7.	a) Expand $\sin x$ in powers of $(x - \frac{f}{2})$.	7M	CO4	L3
	b) Using Maclaurin's series , expand $log(1+x)$ in powers of x .	7M	CO4	L3
8.		14M	CO4	L4
9.	a) Evaluate $\int_{0}^{2} \int_{0}^{3} xy dx dy$	7M	CO5	L3
	b) Evaluate $\int_{0}^{2} \int_{0}^{x} y dy dx$	7M	CO5	L3
10.	a) Evaluate $\int_{0}^{1} x^{5} (1-x)^{3} dx$ using Beta function.	7M	CO5	L3
	b) Evaluate $\int_{0}^{\infty} x^{6} e^{-2x} dx$	7M	CO5	L3

Hall Ticket Number :						

Code: 19AC13T

R-19

I B.Tech. I Semester Supplementary Examinations July 2022

Chemistry of Materials

(Common to CE & ME)

Max. Marks: 70 Time: 3 Hours

Answer any five full questions by choosing one question from each unit (5x14 = 70 Marks)

			Marks	со	Blooms Level
		UNIT-I			
1.		Explain the determination of hardness of water by EDTA method OR	14M	CO1	L3
2.	a)	Define brackish water? What type of method is used for its purification?	7M	CO1	L1
	b)	Describe the treatment of saline water by reverse osmosis in detail	7M	CO1	L2
		UNIT-II			
3.	a)	Describe the working principle of Weston-Cadmium cell with neat diagram	7M	CO2	L4
	b)	Define fuel cell and classify it. List advantages of fuel cell	7M	CO2	L1
		OR			
4.	a)	Define standard electrode potential with example	7M	CO2	L1
	b)	Draw and label Calomel electrode and standard hydrogen electrode	7M	CO2	L4
		UNIT-III			
5.	a)	Discuss the role of inhibitors for the anodic and cathodic protection	7M	CO3	L2
	b)	Write short notes on i) Galvanizing ii) Tanning	7M	CO3	L1
		OR			
6.	a)	Differentiate dry and wet corrosion	7M	CO3	L3
	b)	Write down the equation for the hydrogen evolution type of electrochemical	71.4	000	1.4
		corrosion reaction	/ IVI	CO3	L1
		UNIT-IV			
7.	a)	Differentiate thermosetting and thermoplastic polymers	7M	CO4	L3
	b)	Define polymer with example and classify it	7M	CO4	L1
_		OR			
8.		Explain various steps involves in the manufacturing of Portland cement with a neat labelled diagram of rotary kiln	14M	CO4	L4
9.	a)	What are the uses of smart materials	7M	CO5	L1
	b)	List out the application of nanomaterials		CO5	L3
	-,	OR			_0
10.	a)	Discuss characterization method of nanomaterials	7M	CO5	L2
	b)	List out the applications of TEM		CO5	L3
	,	***			
