#### Code: 20A224T

I B.Tech. II Semester Supplementary Examinations December 2023

# **Electrical Circuits and Technology**

(Electronics and Communication Engineering)

Max. Marks: 70

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Time: 3 Hours

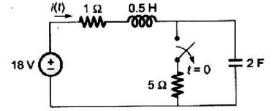
Note: 1. Question Paper consists of two parts (**Part-A** and **Part-B**)

- 2. In Part-A, each question carries **Two marks.**
- 3. Answer ALL the questions in Part-A and Part-B

### PART-A

## (Compulsory question)

- 1. Answer ALL the following short answer questions  $(5 \times 2 = 10M)$  CO BL
- a) Determine the initial conditions of inductor and capacitor shown in the circuit.



- b) Define bandwidth and the impact of resistance on bandwidth.
- c) Express the symmetric and reciprocity condition of a network in terms of transmission parameters. CO3
- d) Mention the merits and demerits of electric braking.
- e) A 3-phase, 50Hz Squirrel cage induction motor runs at 4% slip. What will be the frequency of rotor currents?

#### <u>PART-B</u>

UNIT-I

Answer five questions by choosing one question from each unit ( 5 x 12 = 60 Marks )

Marks CO BL

CO1

CO2

CO4

L3

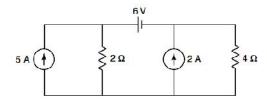
L2

L2

L2

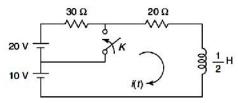
L2

2. a) Find the current in 2 ohm resistor of the Fig. shown below using Source Transformation



6M CO1 L3

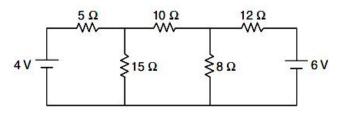
 b) For the given network, a steady state is reached with the switch K open. At t=0, the switch K is closed. Find the current i(t) for t >0.,



6M CO1 L4

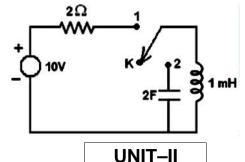
OR

3. a) Find the current through 8 ohm resistor in the circuit shown below using Mesh analysis



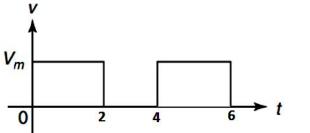
6M CO1 L4

b) For the circuit given in figure steady-state conditions are reached for the switch K in position '1'. At t = 0, the switch is changed to position 2. Use the time domain method to determine the current through the inductor for all t>0.



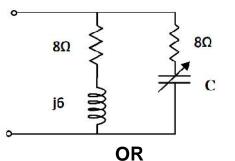
6M CO1 L4

4. a) Find the rms value of the voltage waveform of Figure shown below.



6M CO2 L3

 b) Find C which results in resonance in the circuit shown in Figure when =5000rad/s.



6M CO2 L3

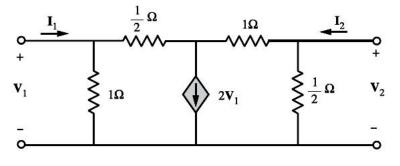
5. a) An RLC series circuit has a current which lags the applied voltage by 45. The voltage across the inductance has a maximum value equal to twice the maximum value of voltage across the capacitor. Voltage across the inductance is  $300\sin(1000t)$  and R = 20 W. Find the value of inductance and capacitance.

6M CO2 L3

b) A series RLC circuit is connected across a variable frequency supply and has R = 12 ohms, L = 1mH and C = 1000PF. Calculate resonant frequency, Q factor and cut of frequencies.

## UNIT–III

- 6. a) Express *h* parameters in terms of admittance parameters for a generalized network.
  - b) Find the transmission parameters for the two-port network shown in Figure.



6M CO3 L4

6M CO2 L3

6M CO3 L2

OR

7. a) Find the h parameters of the network in figure. Invistigate whether the netwok is symmetrical and reciprocal.

 $\begin{array}{c}
5 \Omega \\
6 \Omega \\
4 i_x \\
10 \Omega \\
10 \Omega
\end{array}$ 

b) The following measurements were made on a two-port resistive circuit: With port 1 open, V2 = 15 V, V1 = 10 V, and  $I_2 = 30$  A; with port 1 short-circuited, V2 = 10 V, 12 = 4 and  $I_{1} = -5$  A. Calculate the *z* parameters. 6M CO3 L3

## UNIT–IV

- 8. a) Explain the significance of back EMF of a DC motor. Derive the torque equation of a DC motor.
  - b) A 4-pole, 220V shunt motor has 540 lap-wound conductors. It takes 32 A from the supply mains and develops output power of 5.595 kW. The field winding takes 1 A. The armature resistance is 0.09 and the flux per pole is 30mWb. Calculate (i) the speed and (ii) the torque developed in N-m.

6M CO3 L4

6M CO4 L2

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	ÖK							
9. a)	Draw and explain the speed-torque characteristics DC shunt motor. Write the applications of it.	6M	CO4	L2				
b)	A 250 volt DC shunt motor has armature resistance of 0.25ohm, on load it takes an armature current of 50 A and runs at 750 rpm. If the flux of motor is reduced by 10% without changing the load torque, find the new speed of the motor.	614	CO4					
		OIVI	004	LS				
( <b>a</b> )	UNIT-V							
10. a)	Discuss the effect of variations of frequency and voltage on iron loss. How will you minimize the hysteresis and eddy current loss that occur in a transformer?	6M	CO5	L2				
b)	<ul> <li>b) A 100kVA, 2000/200V, 50Hz single phase transformer has an impedance drop of 10% and resistance drop of 5%. Calculate the (i) regulation at full load 0.8pf lagging and (ii) value of pf at which regulation is zero.</li> </ul>							
	OR	0101	CO5	LJ				
44 ->								
11. a)	Derive an expression of rotor frequency in terms of main supply frequency and slip.	6M	CO5	L2				
b)	A 3-phase, Slip-ring, induction motor with star-connected rotor has an induced e.m.f of 120 Volts between slip-rings at standstill with normal voltage applied to the stator. The rotor winding has a resistance per phase of 0.3 Ohm and							
	standstill leakage reactance per phase of 1.5 Ohm.	6M	CO5	L3				

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H	all Ticket Number :			
Co	ode: 20A421T	R-20		
	I B.Tech. II Semester Supplementary Examinations December	er 2023		
	Electronic Devices and Circuits			
	(Electronics and Communication Engineering)	T		
M	ax. Marks: 70 ********	Time: 3 Hc	Urs	
No	<ul> <li>te: 1. Question Paper consists of two parts (Part-A and Part-B)</li> <li>2. In Part-A, each question carries Two marks.</li> <li>3. Answer ALL the questions in Part-A and Part-B</li> </ul>			
	PART-A			
	(Compulsory question)			
	nswer ALL the following short answer questions $(5 \times 2 =$	10M)	CO	BL
,	xplain thermal stability.		CO1	L1
,	hy we call BJT as a current Controlled Device?			L2
	/hat is thermal runaway?		CO3	L3
•	hat is pinch-off voltage? Give its expression			L2
e) W	hat is meant by Intrinsic standoff ratio and what is its typical va	lue?	CO5	L1
,	PART-B	- 60 Morko	`	
F	Answer <i>five</i> questions by choosing one question from each unit ( 5 x 12	Marks	-	BL
	UNIT–I	marite	00	00
2. a)	With neat circuit explain how transistor acts as an amplifier	6M	1	L1
b)	With neat circuit diagram, explain the principle of self-biasi	ng		
,	of BJT.	6M	2	L2
	OR			
3. a)	In a Silicon transistor circuit with a fixed bias, VCC=9	₽V,		
	RC=3K , RB=8K , =50, VBE=0.7V. Find the operating po	oint		L1,
	and Stability factor.	6M	3	
b)	What is the necessity of biasing circuits? Derive t	he		
	expression for stability factor of self-bias circuit.	6M	2	L2
4. a)	Explain the construction, operation and characteristic behav	ior		
	of JFET under various biasing conditions.	6M	3	L3
b)	Draw and explain small-signal model of a MOSFET.	6M	4	L2
	OR			
5.	Explain the constructional features of a Depletion mode	P-		
	channel and Enhancement mode MOSFET and explain			
	basic operation.	12M	3	L4
		Page <b>1</b>	of <b>2</b>	

		UNIT–III			
6.	a)	Explain about power amplifiers and their features.	6M	2	L4
	b)	Derive an expression for efficiency of class-A amplifier.	6M	3	L5
		OR			
7.	a)	Compare between Class-A, Class-B and Class-C amplifiers.	6M	3	L2
	b)	What are h-parameters, explain how they can be determined from BJT characteristics.	6M	4	L2
		UNIT–IV			
8.		Derive the voltage gain, input admittance and input miller capacitance of CS amplifier using its high frequency equivalent circuit.	12M	5	L3
		OR			
9.		Draw the circuit of source follower Amplifier and derive the expressions for AI, AV, Ri and Ro.	12M	4	L4
10.	a)	With neat diagram explain the construction, working characteristics of UJT. Give its equivalent circuit.	6M	5	L5
	b)	With the help of V-I characteristics, explain SCR operation.	6M	4	L2
		OR			
11.		Define tunneling phenomenon. Explain how tunnel diode operates under different operating conditions. In what way it is different from conventional diodes? Give the necessary energy level diagrams.	12M	5	L4
				0	<b>L</b> T

\*\*\* End \*\*\*

<ul> <li>cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li><b>OR</b></li> <li>5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.</li> <li><b>UNIT-III</b></li> </ul>	Ha	Il Ticket Number :			
I B.Tech. II Semester Supplementary Examinations December 2023         Chemistry         (Common to EEE, ECE, and AI&ML )         Max. Marks: 70       Time: 3 Hours         Note: 1. Question Paper consists of two parts (Part-A and Part-B)         2. In Part-A, each question carries Two marks.       3. Answer ALL the questions in Part-A and Part-B         A (Compulsory question)         1. Answer ALL the following short answer questions (5 X 2 = 10M)       CO       BL         a) Define electrolytic cell. Give any two examples.       CO1       L1         b) What are primary batteries? Give two examples.       CO2       L1         c) What are stereospecific polymers? Give examples.       CO3       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         Marks CO BL         QUNIT-I         2.         Describe the construction and working of a Calomel       electrode with neat sketch.       Marks       C	Cor		R-20		
(Common to EEE, ECE, and AI&ML.)         Max. Marks: 70       Time: 3 Hours         Note: 1. Question Paper consists of two parts (Part-A and Part-B)       2. In Part-A, each question carries Two marks.         3. Answer ALL the questions in Part-A and Part-B       PART-A         (Compulsory question)       1. Answer ALL the following short answer questions (5 X 2 = 10M)       CO       BL         a) Define electrolytic cell. Give any two examples.       CO1       L1         b) What are primary batteries? Give two examples.       CO2       L1         c) What are stereospecific polymers? Give examples.       CO3       L1         d) Define electromagnetic spectrum.       CO4       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) Bescribe the construction and working of a Calomel electrode with neat sketch.       GM       CO1       L2         b) Discuss about polymer membrane electrode and give i	000		2023		
Max. Marks: 70       Time: 3 Hours         Note: 1. Question Paper consists of two parts (Part-A and Part-B)       2. In Part-A, each question carries Two marks.         3. Answer ALL the questions in Part-A and Part-B       PART-A         (Compulsory question)       1. Answer ALL the following short answer questions (5 X 2 = 10M)       CO       BL         a) Define electrolytic cell. Give any two examples.       C01       L1         b) What are primary batteries? Give two examples.       C02       L1         c) What are stereospecific polymers? Give examples.       C03       L1         d) Define electromagnetic spectrum.       C04       L1         e) What are molecular switches? Give two examples.       C05       L1         e) What are molecular switches? Give two examples.       C05       L1         e) What are molecular switches? Give two examples.       C05       L1         e) What are molecular switches? Give two examples.       C05       L1         e) What are molecular sketch.       Marks       C0       BL         2. a) Describe the construction and working of a Calomel electrode with neat sketch.       6M       C01       L2         b) Discuss about polymer membrane electrode and give its significance.       6M       C01       L4         b) Discuss briefly about solid state ion-selective electrodes.       6M<		•			
Note: 1. Question Paper consists of two parts (Part-A and Part-B)         2. In Part-A, each question carries Two marks.         3. Answer ALL the questions in Part-A and Part-B         PART-A         (Compulsory question)         1. Answer ALL the following short answer questions (5 X 2 = 10M)       CO         a) Define electrolytic cell. Give any two examples.       CO1         b) What are primary batteries? Give two examples.       CO2         c) What are stereospecific polymers? Give examples.       CO3         c) Define electromagnetic spectrum.       CO4         c) What are molecular switches? Give two examples.       CO5         c) What are molecular switches? Give two examples.       CO5         LI       What are molecular switches? Give two examples.       CO5         c) Describe the construction and working of a Calomel electrode with neat sketch.       Marks       CO         LI       UNIT-I       Marks       CO       L2         COR       CO1       L4         L       UNIT-II       Marks       GM       CO1       L4         L       UNIT-II       Marks       GM       CO1       L4         L       Discuss about polymer membrane electrode and give its significance.       GM       CO1       L4         L       U	Ma		ime: 3 H	ours	
<ul> <li>2. In Part-A, each question carries Two marks.</li> <li>3. Answer ALL the questions in Part-A and Part-B PART-A (Compulsory question) <ol> <li>Answer ALL the following short answer questions (5 X 2 = 10M)</li> <li>CO BL</li> <li>a) Define electrolytic cell. Give any two examples.</li> <li>CO1 L1</li> <li>b) What are primary batteries? Give two examples.</li> <li>CO3 L1</li> <li>c) What are stereospecific polymers? Give examples.</li> <li>CO4 L1</li> <li>c) What are nolecular switches? Give two examples.</li> <li>CO5 L1</li> <li>d) Define electromagnetic spectrum.</li> <li>c) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>c) What are nolecular switches? Give two examples.</li> <li>CO5 L1</li> <li>c) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>c) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>c) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>c) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>c) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>c) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>c) UNIT-I</li> </ol></li></ul> 2. a) Describe the construction and working of a Calomel electrode with neat sketch. <ul> <li>Marks CO BL</li> <li>Marks CO BL</li> <li>Discuss about polymer membrane electrode and give its significance.</li> <li>An Discuss about polymer membrane electrode and give its significance.</li> <li>CO1 L4</li> <li>b) Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li>COR</li> </ul> 5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications. 12M CO2 L2 UNIT-II		******		0010	
(Compulsory question)         1. Answer ALL the following short answer questions (5 X 2 = 10M)       CO       BL         a) Define electrolytic cell. Give any two examples.       CO1       L1         b) What are primary batteries? Give two examples.       CO2       L1         c) What are stereospecific polymers? Give examples.       CO3       L1         d) Define electromagnetic spectrum.       CO4       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) What are molecular switches? Give two examples.       CO5       L1         e) Wint—I       Q       Q       CO1       L2         b) Discuss about polymer membrane electrode and give its significance.       GM       CO1       L4	Note	2. In Part-A, each question carries Two marks.			
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<ul> <li>b) What are primary batteries? Give two examples.</li> <li>c) What are primary batteries? Give two examples.</li> <li>c) What are stereospecific polymers? Give examples.</li> <li>d) Define electromagnetic spectrum.</li> <li>e) What are molecular switches? Give two examples.</li> <li>CO3 L1</li> <li>d) Define electromagnetic spectrum.</li> <li>c) L1</li> <li>PART-B</li> <li>Answer five questions by choosing one question from each unit (5 x 12 = 60 Marks)</li> <li>Marks CO BL</li> <li>Marks CO BL</li> <li>2. a) Describe the construction and working of a Calomel electrode with neat sketch.</li> <li>b) Distinguish between electrochemical cell and electrolytic cell.</li> <li>6M CO1 L2</li> <li>OR</li> <li>3. a) Discuss about polymer membrane electrode and give its significance.</li> <li>6M CO1 L4</li> <li>Discuss briefly about solid state ion-selective electrodes.</li> <li>6M CO1 L4</li> <li>Answer five questions involved in Zinc- air battery. Give its challenges and applications.</li> <li>CO2 L4</li> <li>OR</li> <li>5. Describe the working principle and construction of propane-oxygen fuel cell with neat sketch. Give its applications.</li> <li>12M CO2 L4</li> <li>UNIT-II</li> </ul>	1.	Answer ALL the following short answer questions (5 $X$ 2 = 10M	, ,	-	L
<ul> <li>c) What are stereospecific polymers? Give examples.</li> <li>c) What are stereospecific polymers? Give examples.</li> <li>d) Define electromagnetic spectrum.</li> <li>e) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>PART-B</li> <li>Answer five questions by choosing one question from each unit (5 x 12 = 60 Marks)</li> <li>Marks CO BL</li> <li>UNIT-I</li> <li>2. a) Describe the construction and working of a Calomel electrode with neat sketch.</li> <li>b) Distinguish between electrochemical cell and electrolytic cell.</li> <li>6M CO1 L2</li> <li>OR</li> <li>3. a) Discuss about polymer membrane electrode and give its significance.</li> <li>6M CO1 L4</li> <li>b) Discuss briefly about solid state ion-selective electrodes.</li> <li>6M CO1 L4</li> <li>b) Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li>12M CO2 L4</li> <li>OR</li> <li>5. Describe the working principle and construction of propane oxygen fuel cell with neat sketch. Give its applications.</li> <li>12M CO2 L2</li> </ul>	a)	Define electrolytic cell. Give any two examples.	C	01 L	1
<ul> <li>d) Define electromagnetic spectrum.</li> <li>e) What are molecular switches? Give two examples.</li> <li>CO4 L1</li> <li>e) What are molecular switches? Give two examples.</li> <li>CO5 L1</li> <li>PART-B</li> <li>Answer five questions by choosing one question from each unit (5 x 12 = 60 Marks)</li> <li>Marks CO BL</li> <li>UNIT-I</li> <li>2. a) Describe the construction and working of a Calomel electrode with neat sketch.</li> <li>b) Distinguish between electrochemical cell and electrolytic cell.</li> <li>6M CO1 L2</li> <li>b) Discuss about polymer membrane electrode and give its significance.</li> <li>6M CO1 L4</li> <li>b) Discuss briefly about solid state ion-selective electrodes.</li> <li>6M CO1 L4</li> <li>Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li>12M CO2 L4</li> <li>OR</li> <li>5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.</li> <li>12M CO2 L2</li> </ul>	b) '	What are primary batteries? Give two examples.	C	02 L	1
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PART-B         Answer five questions by choosing one question from each unit (5 x 12 = 60 Marks)         Marks CO BL         UNIT-I         2. a) Describe the construction and working of a Calomel electrode with neat sketch.       6M CO1 L2         b) Distinguish between electrochemical cell and electrolytic cell.       6M CO1 L2         OR         3. a) Discuss about polymer membrane electrode and give its significance.         BM CO1 L4         b) Discuss briefly about solid state ion-selective electrodes.         OR         A. Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.         OR         5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.         12M CO2 L2         UNIT-II	d)	Define electromagnetic spectrum.	C	D4 L	1
Answer five questions by choosing one question from each unit (5 x 12 = 60 Marks) Marks CO BL UNIT-I 2. a) Describe the construction and working of a Calomel electrode with neat sketch. 6M CO1 L2 b) Distinguish between electrochemical cell and electrolytic cell. 6M CO1 L2 OR 3. a) Discuss about polymer membrane electrode and give its significance. 6M CO1 L4 b) Discuss briefly about solid state ion-selective electrodes. 6M CO1 L4 b) Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications. 12M CO2 L4 OR 5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications. 12M CO2 L2	e)	What are molecular switches? Give two examples.	C	D5 L	1
UNIT-I         2. a) Describe the construction and working of a Calomel electrode with neat sketch.       6M CO1 L2         b) Distinguish between electrochemical cell and electrolytic cell.       6M CO1 L2         b) Discuss about polymer membrane electrode and give its significance.       6M CO1 L4         b) Discuss briefly about solid state ion-selective electrodes.       6M CO1 L4         b) Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.       12M CO2 L4         OR         5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.         12M CO2 L2					
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<ul> <li>2. a) Describe the construction and working of a Calomel electrode with neat sketch.</li> <li>b) Distinguish between electrochemical cell and electrolytic cell.</li> <li>b) Discuss about polymer membrane electrode and give its significance.</li> <li>b) Discuss briefly about solid state ion-selective electrodes.</li> <li>c) CN</li> <li>6M CO1 L2</li> <li>6M CO1 L2</li> <li>6M CO1 L4</li> <li>b) Discuss briefly about solid state ion-selective electrodes.</li> <li>6M CO1 L4</li> <li>6M CO1 L4</li> <li>6M CO1 L4</li> <li>6M CO1 L4</li> <li>CO1 L4</li> <li>CO2 L4&lt;</li></ul>			Marks	CO	BL
<ul> <li>electrode with neat sketch.</li> <li>b) Distinguish between electrochemical cell and electrolytic cell.</li> <li>6M CO1 L2</li> <li>6M CO1 L2</li> <li>6M CO1 L2</li> <li>6M CO1 L2</li> <li>6M CO1 L4</li> <li>CO1 L4</li> <li>CO1 E1</li> <li>CO1 CO1 L4</li> <li>CO1 CO1 CO1</li> <li>CO1 CO1</li> <li>CO1</li> <li>CO1 CO1</li> <li>CO1</li> <li>CO2</li> <li>CO2</li> <li>CO2</li> <li>CO1</li> <li>CO2</li> <li>CO2</li> <li>CO1</li> <li>CO2</li> <li>CO3</li> <li>CO3</li> <li>CO3</li> <li>CO3</li> <li>CO4</li> <li>CO4</li></ul>	2 a)				
<ul> <li>b) Distinguish between electrochemical cell and electrolytic cell.</li> <li>b) Distinguish between electrochemical cell and electrolytic cell.</li> <li>COR</li> <li>CO1 L2</li> <li>OR</li> <li>CO1 L4</li> <li>Discuss briefly about solid state ion-selective electrodes.</li> <li>CO1 L4</li> <li>Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li>COR</li> <li>Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.</li> <li>CO2 L2</li> <li>UNIT-II</li> </ul>	z. uj	•	6M	CO1	12
OR 3. a) Discuss about polymer membrane electrode and give its significance. 6M CO1 L4 b) Discuss briefly about solid state ion-selective electrodes. 6M CO1 L4 UNIT-II 4. Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications. 12M CO2 L4 OR 5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications. 12M CO2 L2	b)	Distinguish between electrochemical cell and electrolytic cell.			
<ul> <li>3. a) Discuss about polymer membrane electrode and give its significance.</li> <li>b) Discuss briefly about solid state ion-selective electrodes.</li> <li>4. Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li>5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.</li> <li>5. UNIT-II</li> </ul>	,		••••	001	
<ul> <li>significance.</li> <li>b) Discuss briefly about solid state ion-selective electrodes.</li> <li>6M CO1 L4</li> <li>6M CO1 L4</li> <li>CO1 L4</li> <li>CO2 L1</li> <li>CO2 L1</li> <li></li></ul>	3 a)				
<ul> <li>b) Discuss briefly about solid state ion-selective electrodes.</li> <li>4. Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li>5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.</li> <li>6. UNIT-III</li> </ul>	0. uj		6M	CO1	14
<ul> <li>UNIT-II</li> <li>Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li>12M CO2 L4</li> <li>OR</li> <li>Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.</li> <li>12M CO2 L2</li> </ul>	b)	0			
<ul> <li>4. Discuss about the chemical composition and the respective cell reactions involved in Zinc- air battery. Give its challenges and applications. 12M CO2 L4</li> <li><b>OR</b></li> <li>5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications. 12M CO2 L2</li> </ul>	~)		0	001	64
<ul> <li>cell reactions involved in Zinc- air battery. Give its challenges and applications.</li> <li>Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications.</li> <li>UNIT-III</li> </ul>	4.				
OR 5. Describe the working principle and construction of propane -oxygen fuel cell with neat sketch. Give its applications. 12M co2 L2 UNIT-III					
<ul> <li>Describe the working principle and construction of propane</li> <li>-oxygen fuel cell with neat sketch. Give its applications.</li> <li>12M CO2 L2</li> <li>UNIT-III</li> </ul>		challenges and applications.	12M	CO2	L4
-oxygen fuel cell with neat sketch. Give its applications. 12M CO2 L2		OR			
UNIT–III	5.	Describe the working principle and construction of propane			
		-oxygen fuel cell with neat sketch. Give its applications.	12M	CO2	L2
2. Discuss shout the detailed assumed for assumed in a set		UNIT–III			
	6.	Discuss about the detailed preparation, properties, and			
applications of Bakelite. 12M CO3 L2		applications of Bakelite.	12M	CO3	L2

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OR
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7. a)	Discuss about coordination addition polymerization (Ziegler -Natta Catalysis) with suitable example.	6M	CO3	L4
b)	Give the preparation and applications of the following polymers. i) Nylon-6 ii) Buna-S	6M	CO3	L2
8. a)	Discuss about the principle and working of a gas chromatography with suitable example.	6M	CO4	L4
b)	Describe the working principle and applications of UV-			
,	Visible spectroscopy.	6M	CO4	L2
	OR			
9.	Write short note on i) pH metry			
	ii) Potentiometry			
	· · ·	4014		
	iii) Conductometry	12IVI	CO4	L2
	UNIT-V			
10. a)	What are artificial molecular machines? Explain with			
	suitable examples.	6M	CO5	L2
b)	Discuss about an autonomous light-powered molecular			
	motor.	6M	CO5	L4
	OR			
11. a)	Discuss about the systems based on Catenanes	6M	CO5	L4
b)	Discuss about the acid-base controlled molecular shuttle.	6M	CO5	L4
0)	*** End ***		000	L4

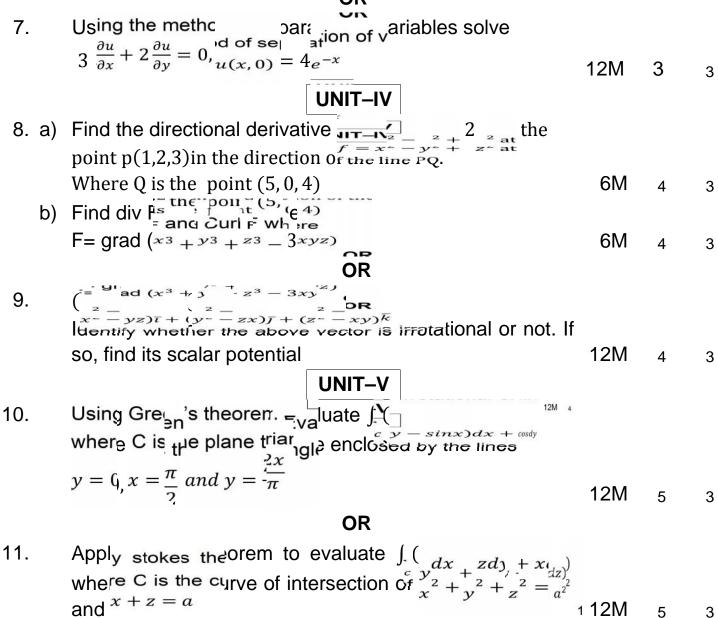
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	N	lote: 1. Question Paper	con	into	of tu		****		<b>A</b> or	d D	art I	2)					
	1	2. In Part-A, each				-					ai t-1	<b>)</b> )					
		3. Answer <b>ALL</b> th	-							-B							
			_			]	PAR'	T-A									
					(C	ompi	ulsor	y que	estior	I)							
	1. <b>A</b>	nswer ALL the follow	ing s	hort	ans	wer	ques	tion	s (5	5 X 2	2 = 10	) ( M			C	O BL	-
	a)	What is Hazlitt's reasor	whe	n he	sen	ds hi	s son	n to s	choo	?					СС	D1 L2	2
	b)	Which lines are repeate	ed in t	the p	oem	"The	e Bro	ok"?	Wha	t do	they	mea	ın?		СС	D1 L2	2
	c)	What are the two dynas	sties i	ment	ione	d in t	he st	ory "	The	Deat	h Tra	ap"?			СС	D1 L2	2
	d)	Explain the attitude of Y	′unus	s' mo	other	towa	rds t	he po	oor a	nd th	ne ne	edy.			СС	D1 L2	2
	e)	What are the two chara	acter	istics	s of r	node	ern la	ngua	age u	sage	es th	at ar	e despi	sed by			
		George Orwell?													CC	D1 L2	<u>'</u>
							PAR		-			•	/= / •	~~			
		Answer <i>five</i> question	ns by	chc	osir	ng or	ne qu	lesti	on fr	om	each	unit	: (5 x 12				
														Ma	rks	CO	BL
2.		Do you think that the	icoc	a of a	advia				liam	⊔o∍l	itt or	o otill	rolovor	*2			
Ζ.		Do you think that the p If yes, explain their rel				e giv	en b	y vvii	llam	nazi	ill an	e Suii	releval		2M	CO3	L4
							OF	2									
3.	a)	Change the following	state	ment	ts inte	o que											
	,	i. Students are lea															
		ii. The tiger mauled	the	deer													
		iii. He has spent a l	ot of	time	work	king d	on his	s con	npute	er.							
		iv. We won accolad			•		ance.										
		v. The dog was pla															
		vi. We have been g	•		•			•							6M	CO3	L3
	b)	Identify the parts of sp					ined	word	is in t	he fo	ollow	ing s	entence	es.			
		i. I ran fast <u>but</u> I <u>mi</u> ii. <u>Barrowed</u> garme															
		iii. He is <u>afraid</u> of <u>fire</u>			III. <u>vv</u>	<u>on</u> .									6M	CO3	L3
							UNIT	'-II									
4.		Develop the following	hints	into	a we	ell - c	onstr	ucte	d par	agra	ph:						
		Regular exercise – be	enefit	s bo	th ph	nysica	ally a	and n	nenta	ully –	· imp	rove	s health	ı —			
		running, swimming a	-		-	-							-	•			
		management – burns										•	•				
		stress – releases en insomnia – long-term	•										s – cur		2M	CO2	L4
							0				.,						_ ·

OR

		OR			
5.		Describe the brook's journey before it flows through Phillip's farm.	12M	CO2	L4
0			4014	000	
6.		Justify the title of Saki's play "The Death Trap".	12IVI	CO3	L4
_		OR			
7.	a)	<ul> <li>Rearrange the following jumbled sentences below so as to have a well- written paragraph.</li> <li>i. We should make sure we get this right.</li> <li>ii. Second, it is crucial that a startup has a great team.</li> <li>iii. Many people may be able to spin out great ideas but it is essential to have clarity on who would pay for our product and why.</li> <li>iv. Finally, let's not forget about the funds. Money is the key factor because without it we can't run a business.</li> <li>v. The success of a startup depends upon a committed, capable and experienced team.</li> <li>vi. To sum up, a good product is the foundation on which we can form a successful business using the funds procured from various sources.</li> <li>vii. Firstly, it is important to have a viable product.</li> </ul>	7M	CO4	13
	<b>۲</b>		<i>i</i> IVI	004	L3
	b)	Put the verbs in brackets in the correct form. i. I never skiing. (try) ii. Patel that historians have misinterpreted the period. (say) iii. Sohail the silver medal in the sprint last year. (win) iv. Dad can't answer the phone right now; he (drive) v. If I early this morning, I would not have missed the bus. (wake up)	5M	CO3	L3
_		UNIT-IV			
8.		How did the Grameen Bank help the impoverished women of Bangladesh? <b>OR</b>	12M	CO4	L4
9.		Prepare an analytical essay on the topic, "The Impact of Social Media in the Society."	12M	CO4	L4
		UNIT–V			
10.		<ul> <li>Correct the following sentences:</li> <li>a) He climbed across the wall and ran until the main road.</li> <li>b) Padma congratulated Tenzin for his promotion.</li> <li>c) We must try harder to stop these people from destroying the nature.</li> <li>d) Working in the lab for hours, my feet began to hurt.</li> <li>e) One of my friend is coming today.</li> <li>f) The people of the district are happy. Because a Medical College will be set up here in a year.</li> <li>g) When I will arrive at the airport, I will call you.</li> <li>h) Let us discuss about tomorrow's programme.</li> <li>i) She returned back home and collapsed on the bed late at night.</li> <li>j) I have gone out yesterday.</li> <li>k) Though the machine was new, but it did not work.</li> <li>l) The bag was very heavy that I could not carry it.</li> </ul>	12M	CO3	L3
11.		What do Orwell's five examples have in common?	12M	CO2	L4

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		Differe	ential							Cal	culu	JS		
	(Common to all Branches) Max. Marks: 70 Time: 3 Hours													
	Note	<ol> <li>1. Question Paper</li> <li>2. In Part-A, each</li> <li>3. Answer ALL th</li> </ol>	questic	n carr	ies <b>Two</b> i n <b>Part-A</b>	mark	s.		art-B	3)				
				( <b>C</b>	<u>PAR</u> ompulsor		estiar	n)						
1.	Answe	er ALL the followi	ng sh		-	-			(5)	X 2	= 10	M )	CC	) BL
	a) Fir	nd the particula	r integ	gral o	$f \begin{pmatrix} \frac{1}{u} \\ e^{t} \\ e^{t} \\ D^{2} \\ \end{pmatrix}$		ons + 1	יי ער	(5) 	× 2	= 10		1	1 2
		rite the second											2	2 3
	c) So	olve p-q=1		C				-					3	3 2
	d) Fir	nd curl F at the	point	(1,2,3	3) giver	ר F=		≣ <sup>*</sup> ั่ลน - 2 ง เ	atio	n f <sup>L</sup>	rm ~ T +		2 3 4	43
		ate Gauss Dive					0	242	se	x y 2	. 2 ) न	- XYZZKJ	5	5 3
	,		0		PAR	T-B								
		Answer five question	ons by c	hoosir			n fror	n eao	ch un	it (5	x 12 =			
					UNI	T–I						Marks	CO	BL
	2.	Solve $(p - 2)^2$	v = e	$\left _{e^{2x}}\right $		⊤_ı F R	- 	•				12M	1	3
	3.	Solve $(D^2 + 3)$	D+2	2) y :	$= e^{-x} -$	$+x^2$	2+0	cos	SX			12M	1	3
			•		UN	T–II								
	4.	Solve $\sum_{x^2 \frac{d^2y}{dx^2}}^{L}$ +	$-2_{x}\frac{6}{6}$	$\frac{ly}{lx} = 2$	0	-	log	X				12M	2	3
	_					R								
	5.	An uncharged	con	en Ignise	er of ca	apac	itv	C is	s ch	arg	ed b	у		
		applying an e.												
		L and negligib	le res	istan	ce, prø	ve ti	nat :	at a	г <b>у t</b>	ime	t, th	е		
		charge on one	of the	e pla	tes is $\frac{E^2}{2}$	$-\{si\}$	$n \frac{1}{\sqrt{L_0}}$	<u> </u>	$\sqrt{\frac{1}{LC}}$	cos -	$\left[\frac{1}{LC}\right]$	12M	2	3
					UNI	т–II								
	6. a)	Form the pa	rtial o	differ	ential e	equa	atior	h b	у е	limi	natin	g		
		arbitrary function		-		-			-			6M	3	3
	b)	Identify the ap	propri	ate f	orm and	d so	lve	$(p) + (p^2) = (p^2)^2$	у е 8(x - + 9 <sup>2</sup>	-a	x + y	6M	3	3

### OR



\*\*\* End \*\*\*

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